



# Animal slurry acidification: more than a solution for ammonia emissions abatement?

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# Structure of the presentation

- 1) Introduction**
- 2) Slurry composition and separation**
- 3) Gaseous emissions during storage**
- 4) N, P and C dynamics after soil application**
- 5) Agronomic value**

# Introduction

- ✓ **pH lowering of animal manure: logical and direct solution to minimize ammonia emissions.**



- ✓ **Additives used: nitric and sulfuric acid with liquid manure and aluminium sulfate with solid manure**



# Introduction

- ✓ **Today, safe and efficient solutions are proposed to farmers for slurry acidification in barn, in slurry store or immediatly before soil application**
- ✓ **But such service is still limited to Denmark where 15% of slurry was acidified in 2013 with an expected increase to 20% in 2014**
- ✓ **More information is needed to export such technology to countries from South Europe**



# Introduction

- **Strong focus of research on efficiency to decrease  $\text{NH}_3$  but few data on other gas emissions neither on N, P and C dynamics in soil;**
- **Most studies performed in North Europe and few available information on the applicability of this technique in Mediterranean countries where pedo-climatic conditions are very different from North Europe.**

 **Provide an overview about the potential for slurry acidification application as a slurry management tool in Mediterranean countries.**

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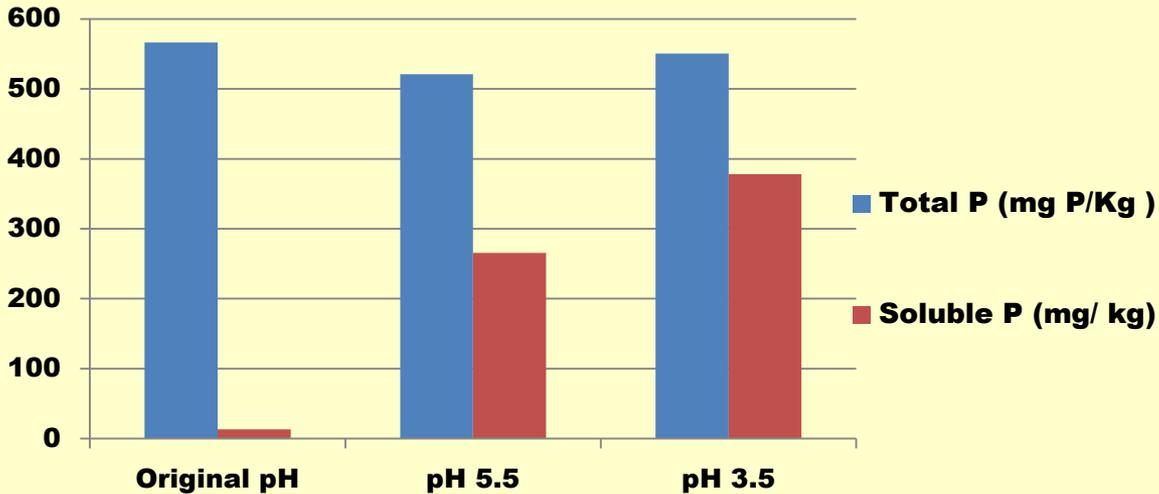
**5) Agronomic value**

# Slurry composition and separation

	<b>Pig slurry</b>	<b>Acidified pig slurry</b>
<b>Dry matter content (g kg<sup>-1</sup>)</b>	<b>49.6</b>	<b>62.6</b>
<b>Total N (g kg<sup>-1</sup>)</b>	<b>4.2</b>	<b>4.3</b>
<b>Total P (g kg<sup>-1</sup>)</b>	<b>1.1</b>	<b>1.0</b>
<b>Total C (g kg<sup>-1</sup>)</b>	<b>14.9</b>	<b>13.3</b>
<b>Inorganic C(g kg<sup>-1</sup>)</b>	<b>0.8</b>	<b>0.0</b>
<b>Ca (g kg<sup>-1</sup>)</b>	<b>2.4</b>	<b>2.4</b>
<b>Mg (g kg<sup>-1</sup>)</b>	<b>1.1</b>	<b>1.0</b>

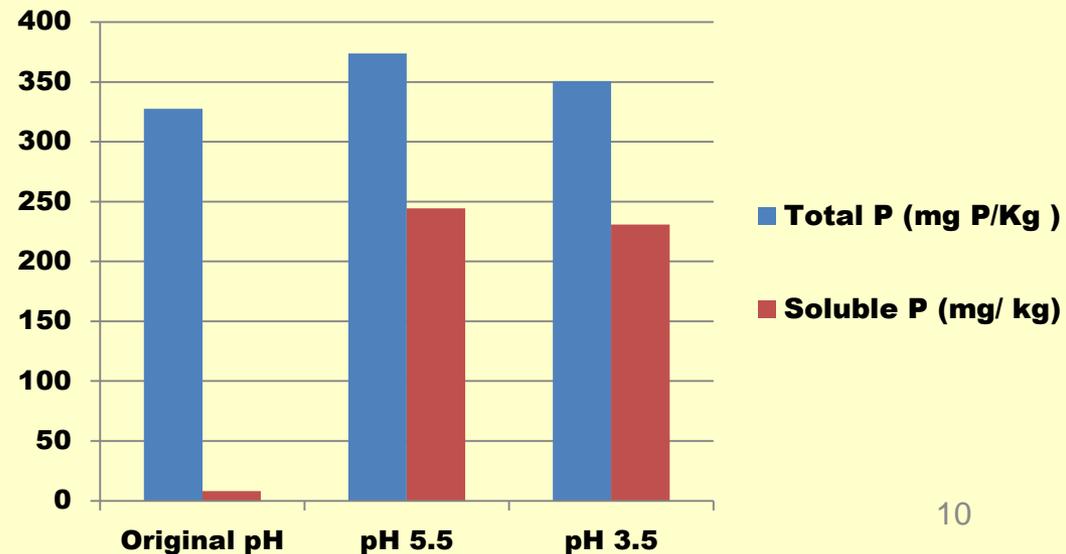
# Slurry composition

## Phosphorous - dairy slurry



**Influence of pH  
target storage time**

## Phosphorous - pig slurry



**Possible dissolution of the  
dominants mineral P  
species in manure (struvite  
and di-calcium phosphate)**

# Slurry composition and separation

	<b>Slurry</b>	<b>Acidified Slurry</b>	<b>Liquid fraction</b>	<b>Acidified Liquid fraction</b>	<b>Solid fraction</b>	<b>Acidified Solid fraction</b>
<b>Dry matter content (g kg<sup>-1</sup>)</b>	<b>49.6</b>	<b>62.6</b>	<b>10.2</b>	<b>30.9</b>	<b>194.8</b>	<b>195.5</b>
<b>Total N (g kg<sup>-1</sup>)</b>	<b>4.2</b>	<b>4.3</b>	<b>2.8</b>	<b>2.9</b>	<b>10.4</b>	<b>9.9</b>
<b>Total P (g kg<sup>-1</sup>)</b>	<b>1.1</b>	<b>1.0</b>	<b>0.04</b>	<b>0.69</b>	<b>4.7</b>	<b>2.1</b>
<b>Total C (g kg<sup>-1</sup>)</b>	<b>14.9</b>	<b>13.3</b>	<b>2.7</b>	<b>1.9</b>	<b>72.4</b>	<b>67.6</b>
<b>Inorganic C (g kg<sup>-1</sup>)</b>	<b>0.8</b>	<b>0.0</b>	<b>0.4</b>	<b>0</b>	<b>2.4</b>	<b>1.0</b>
<b>Ca (g kg<sup>-1</sup>)</b>	<b>2.4</b>	<b>2.4</b>	<b>0.1</b>	<b>0.9</b>	<b>10.6</b>	<b>6.7</b>
<b>Mg (g kg<sup>-1</sup>)</b>	<b>1.1</b>	<b>1.0</b>	<b>0.05</b>	<b>0.5</b>	<b>4.7</b>	<b>2.8</b>

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# Gaseous emissions during storage

## Methane Emissions

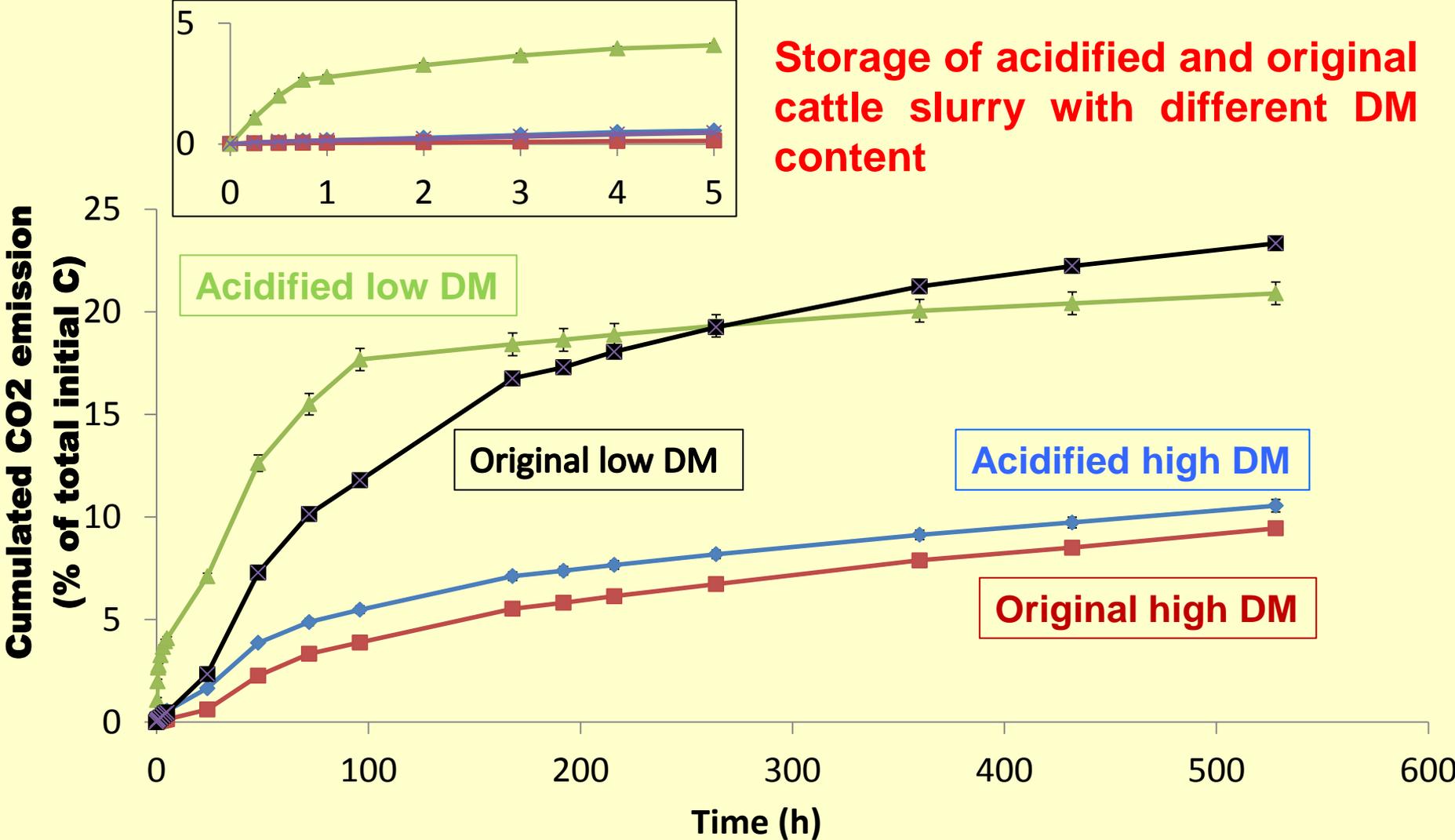
- **Slurry acidification can decrease CH<sub>4</sub> emissions during storage (Berg et al., 2003)**
- **Acidification efficiency to decrease CH<sub>4</sub> emissions depends strongly on the acid used (Berg et al., 2003; Petersen et al.2012) :**
  - **>90% with lactic acid**
  - **40-65 % with HCl**
  - **17-75% with nitric acid**
- **Below pH 5, this decrease does not depends on the target pH .**

# Gaseous emissions during storage

## Carbon dioxide emissions

- ✓ **CO<sub>2</sub> emission occurred mainly during the acidification process and can be 2-10 times higher than during subsequent storage (Fangueiro et al., 2013; Dai and Blanes Vidal, 2013).**
- ✓ **A stronger and faster decay of CO<sub>2</sub> emissions is observed in acidified slurry relative to non acidified during the first days of storage.**
- ✓ **Over the whole storage period, differences between acidified and non acidified slurry in terms of CO<sub>2</sub> emissions were generally not significant.**

# Gaseous emissions during storage



Fangueiro D., Surgy S., Coutinho J., Vasconcelos E. 2012 Impact of cattle slurry acidification on carbon and nitrogen dynamics during storage and after soil incorporation. Journal of Plant Nutrition and Soil Science, In press

# Gaseous emissions during storage

## Hydrogen sulfide emissions

- ✓ **As occurred with CO<sub>2</sub> emissions, a strong burst of H<sub>2</sub>S emissions may happened during the acidification process followed by a strong decrease over the first days of subsequent storage.**
- ✓ **BUT acidification has no significant effect on H<sub>2</sub>S emissions over the whole storage period (Dai and Blanes-Vidal (2012))**
- ✓ **More than the pH effect, the slurry mixing that was performed in all treatments strongly influence H<sub>2</sub>S emissions.**

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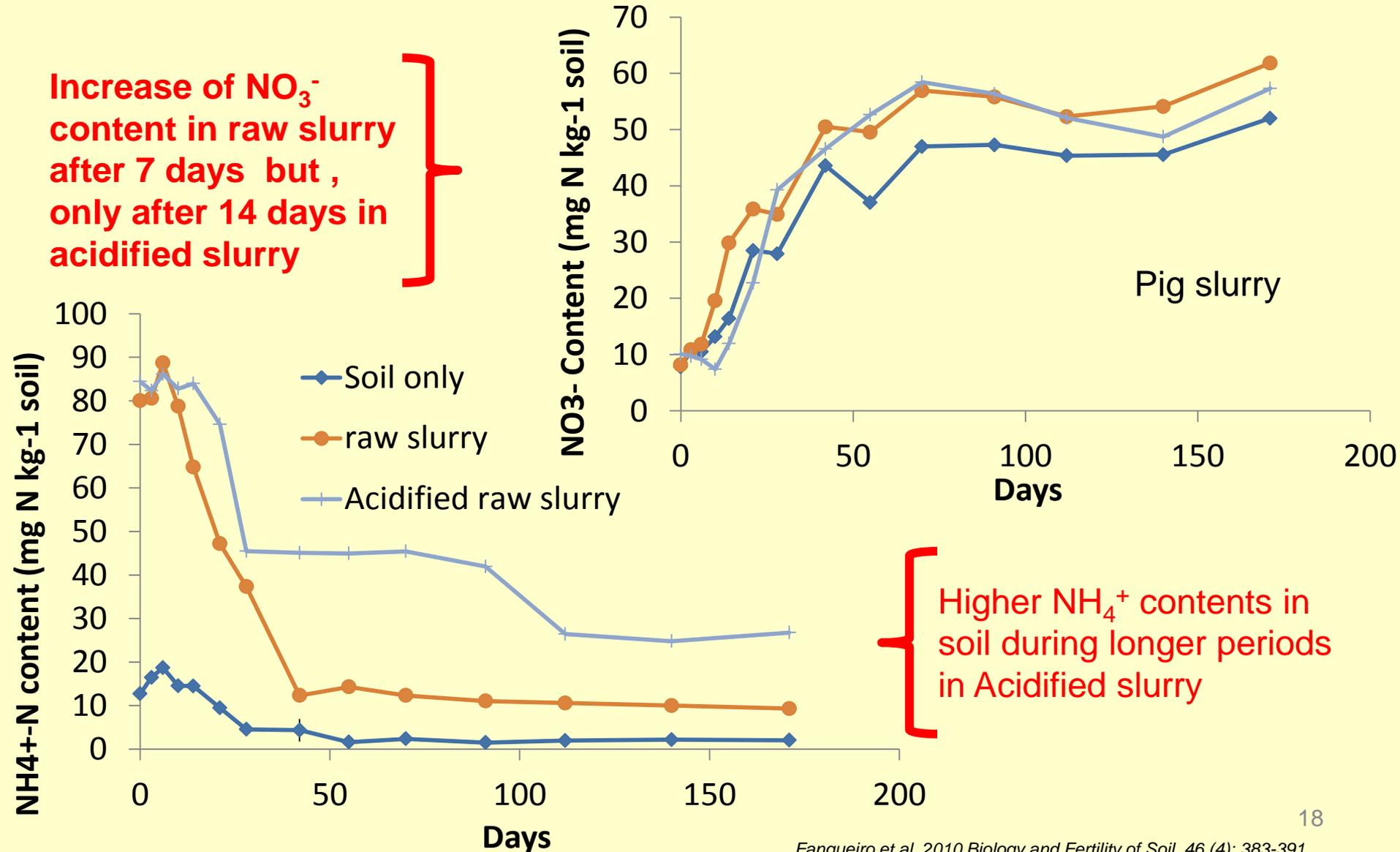
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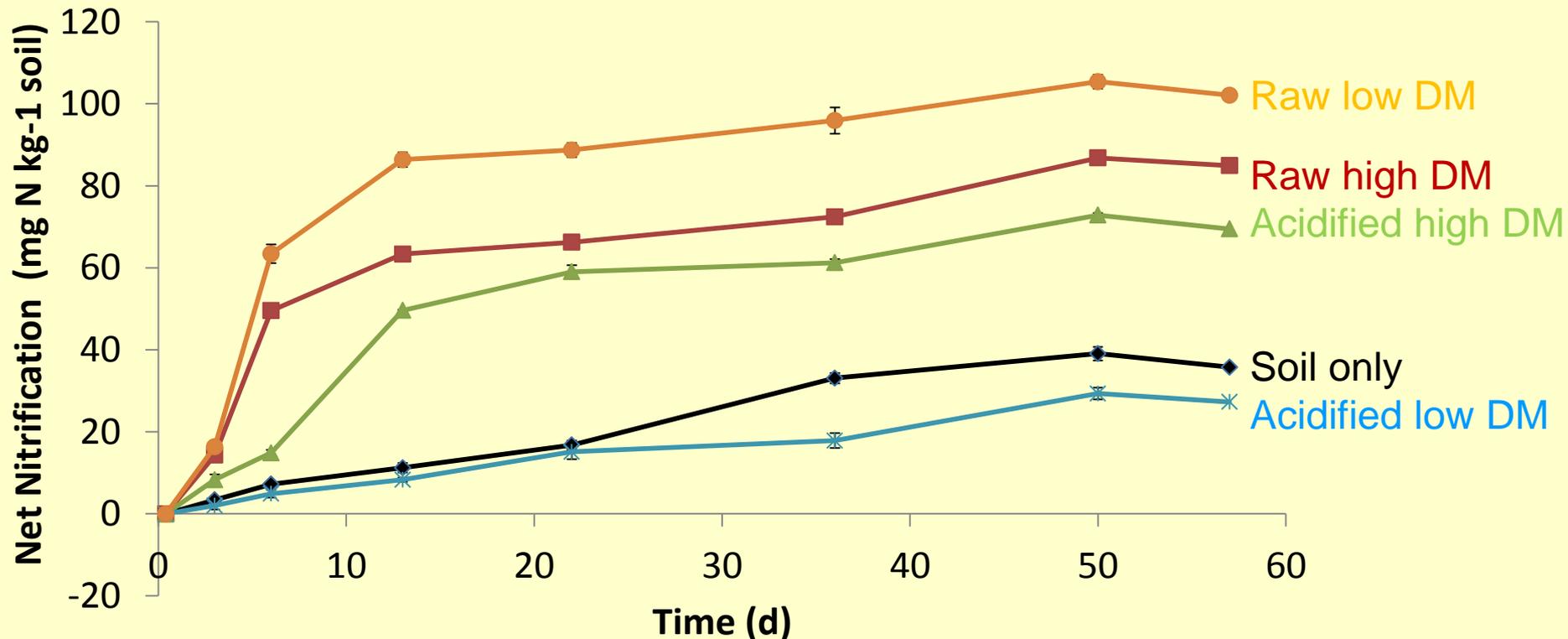
# Influence of slurry acidification on N mineralization and nitrification

Increase of  $\text{NO}_3^-$  content in raw slurry after 7 days but, only after 14 days in acidified slurry



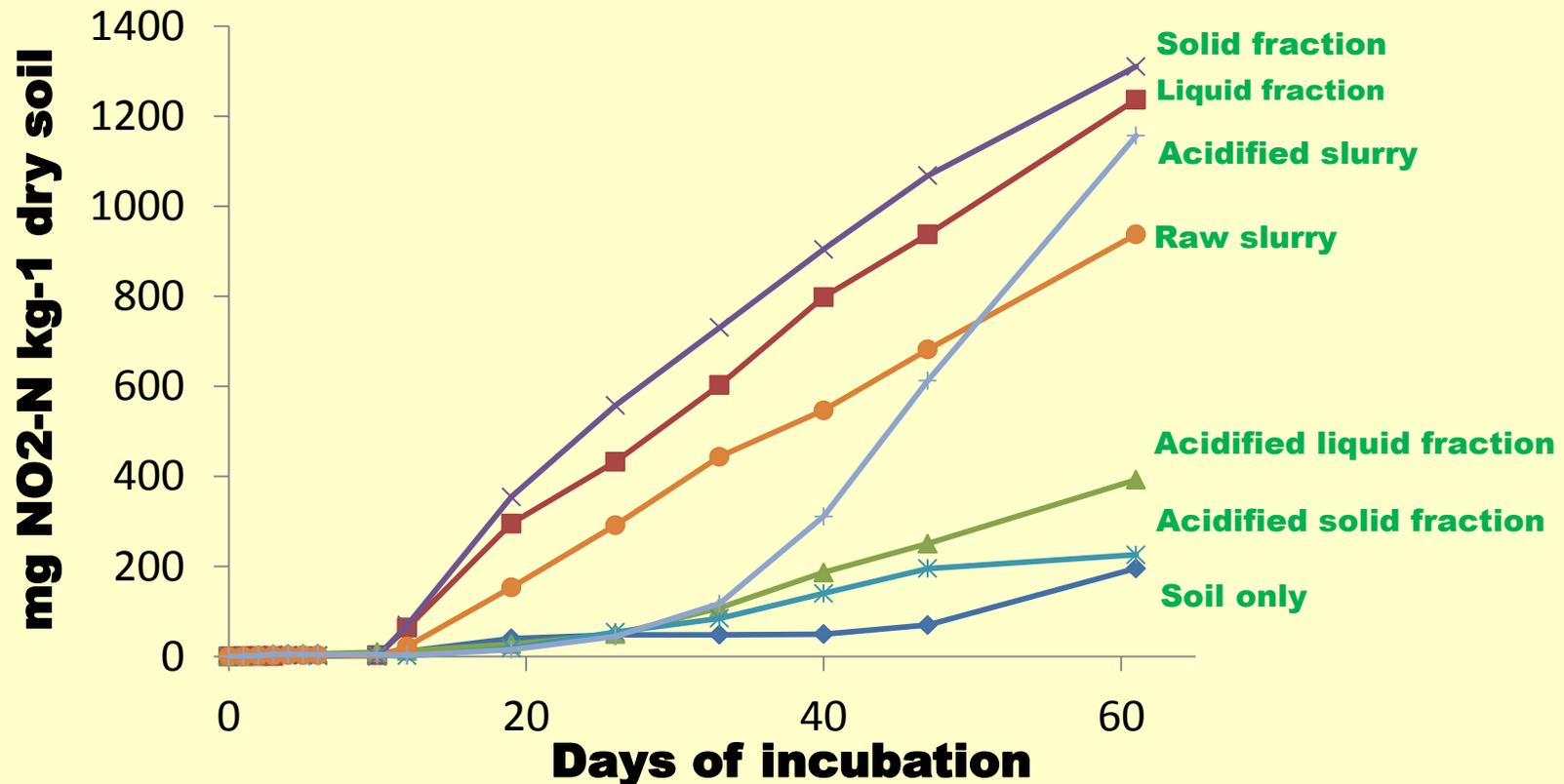
# Influence of slurry acidification on N mineralization and nitrification

**Delay and decrease of nitrification with acidification: effect more intense in low DM slurry**

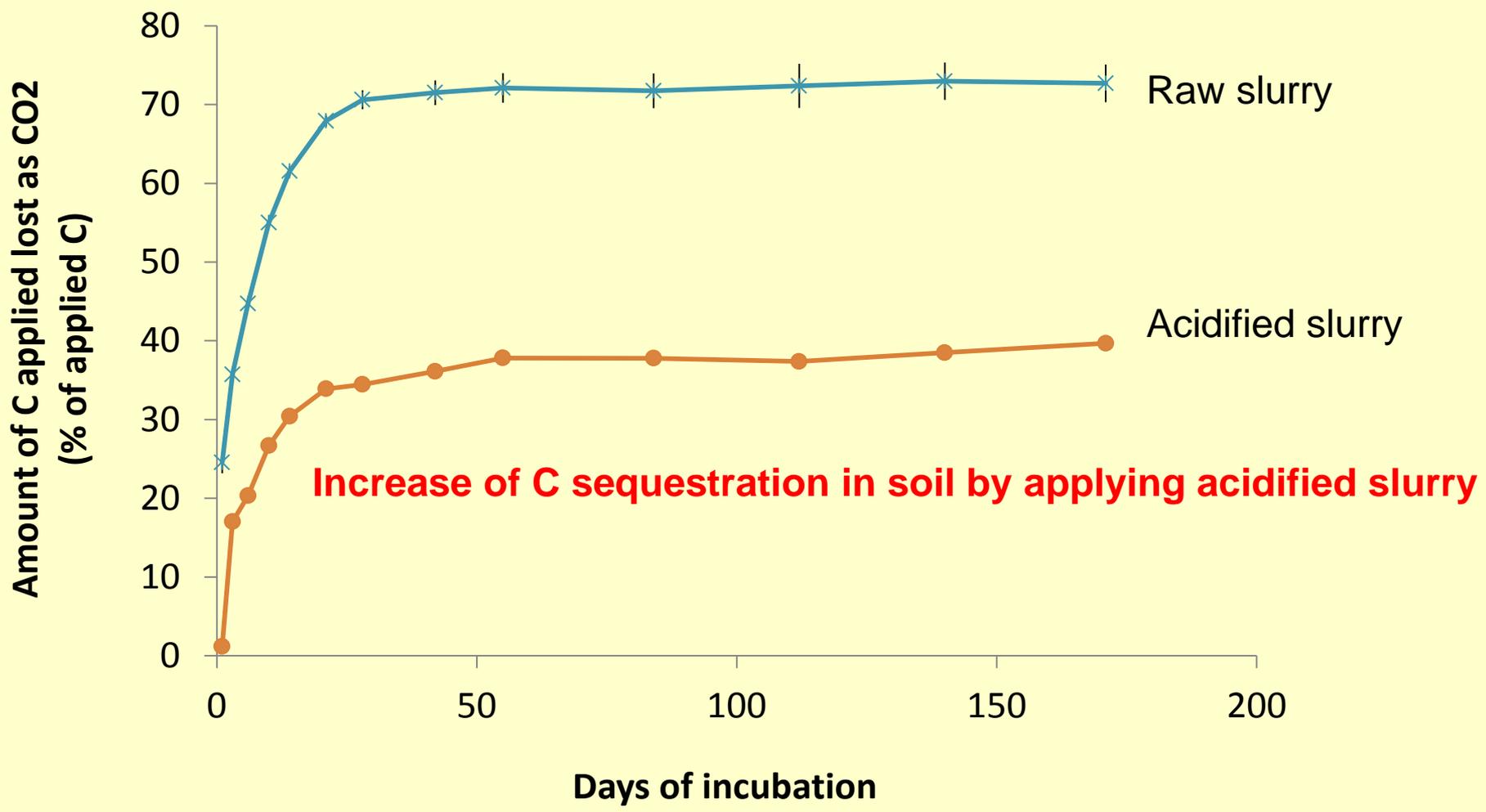


*Soil application of acidified and non acidified cattle slurry with different DM content*

# Influence of slurry acidification on N<sub>2</sub>O after soil application – pig slurry



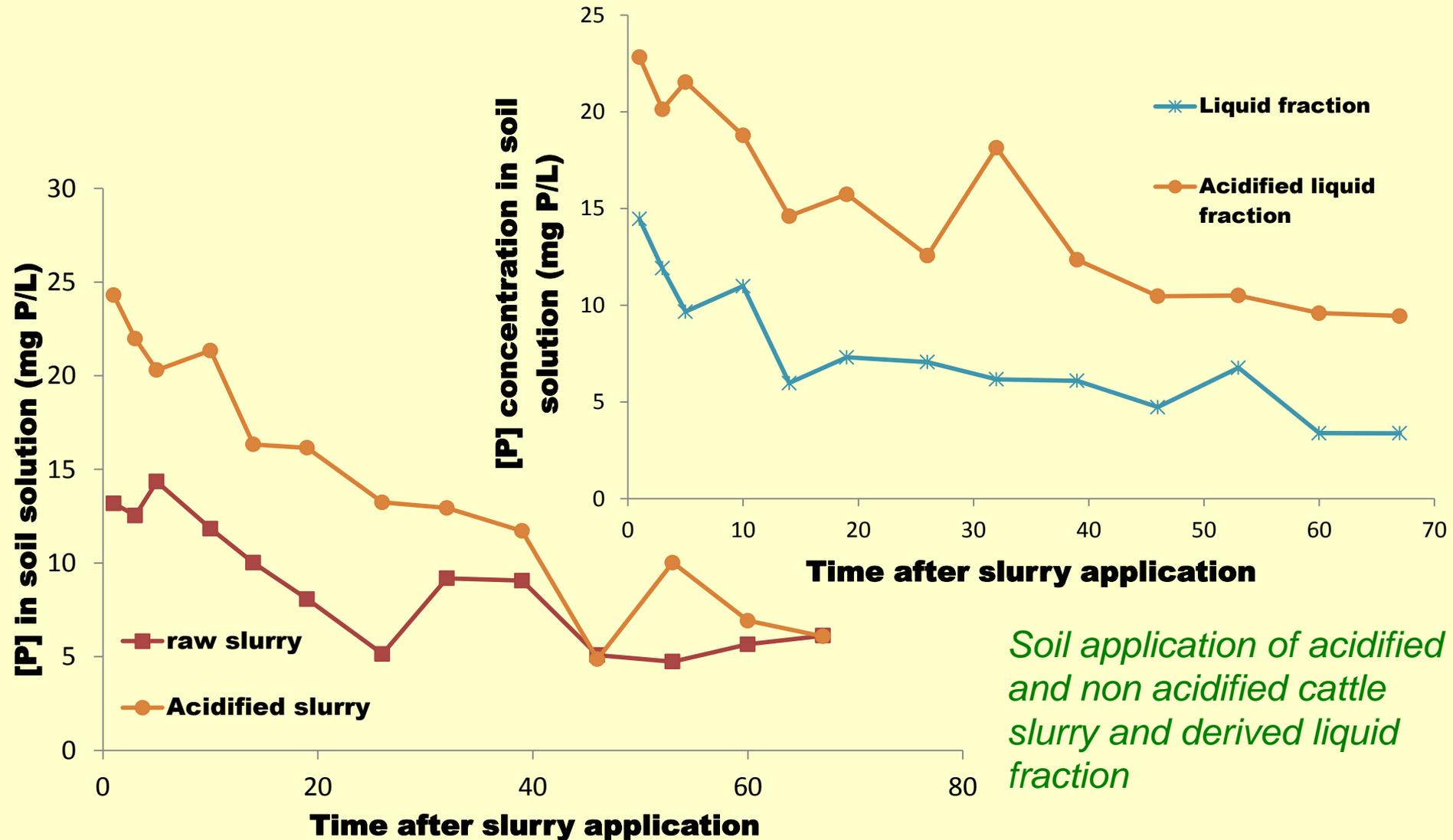
# Influence of slurry acidification on CO<sub>2</sub> emissions after soil application – pig slurry



Fangueiro D., Ribeiro H., Coutinho J., Cardenas L., Trindade H., Cunha-Queda C., Vasconcelos E., Cabral F. 2010 Nitrogen mineralization and CO<sub>2</sub> and N<sub>2</sub>O emissions in a sandy soil amended with original or acidified pig slurries or with the relative fractions. *Biology and Fertility of Soil*, 46 (4): 383-391.

# Influence of slurry acidification on P availability

Slurry acidification increases P availability for plants



Soil application of acidified and non acidified cattle slurry and derived liquid fraction

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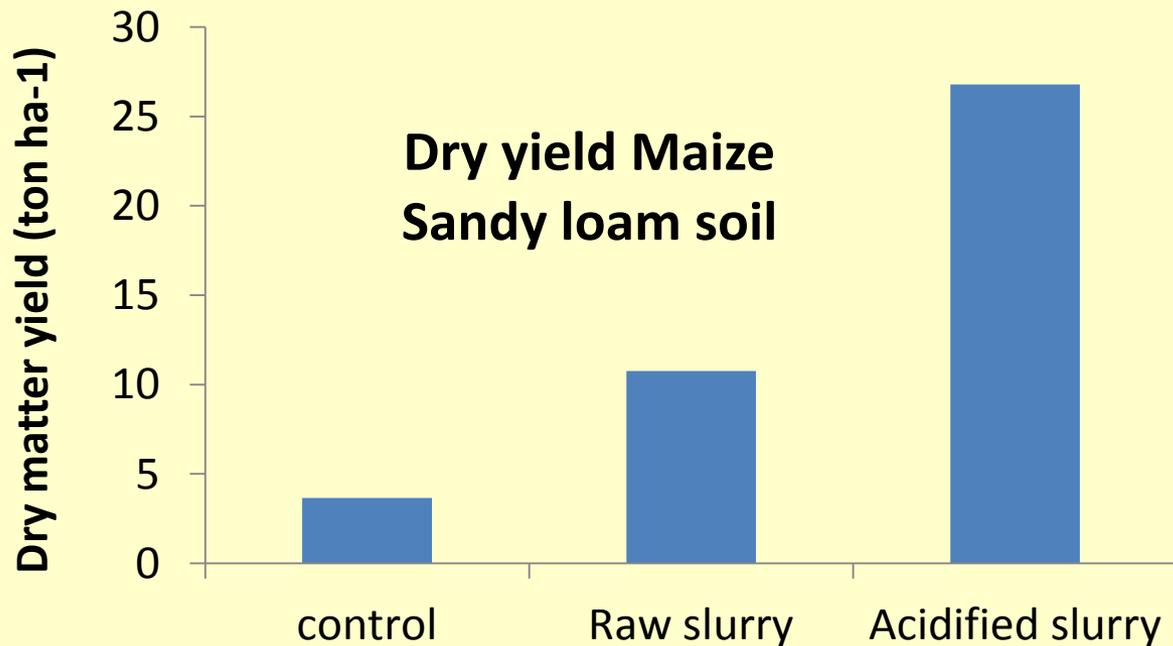
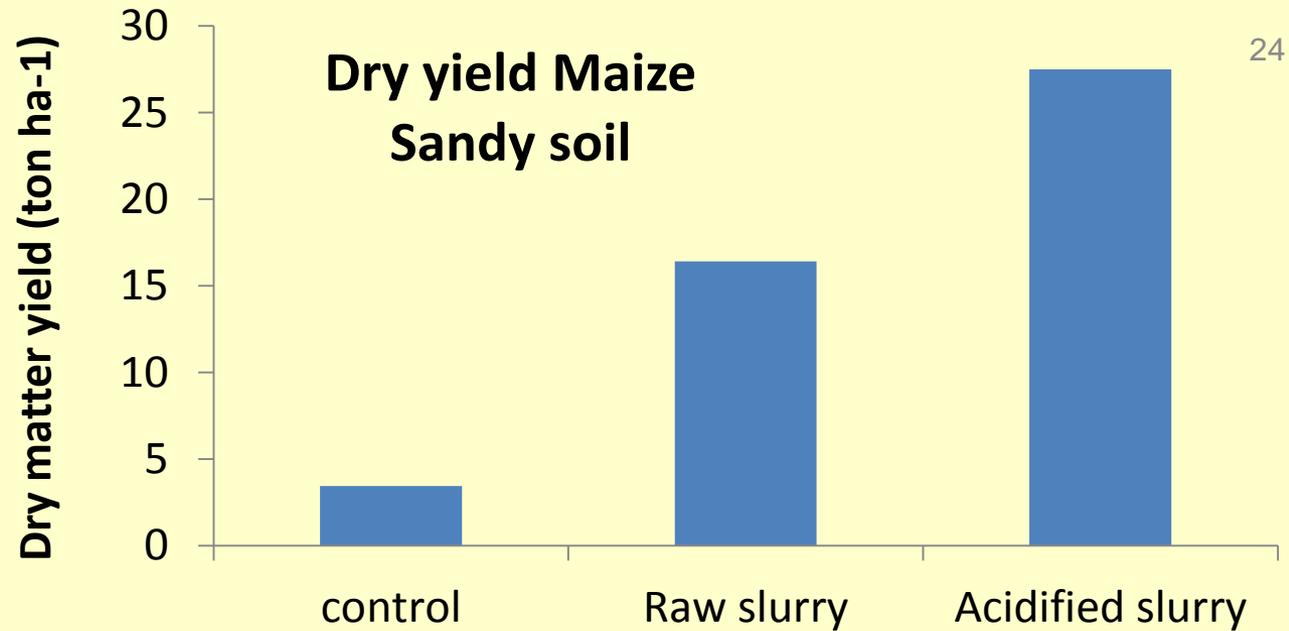
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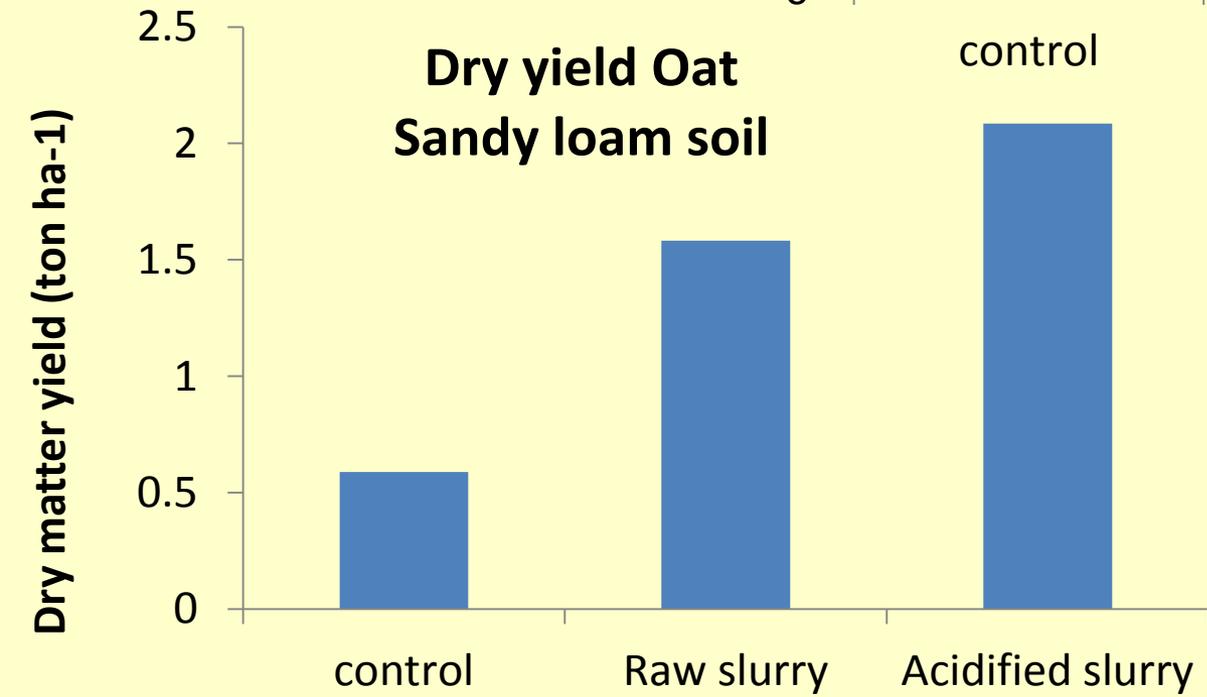
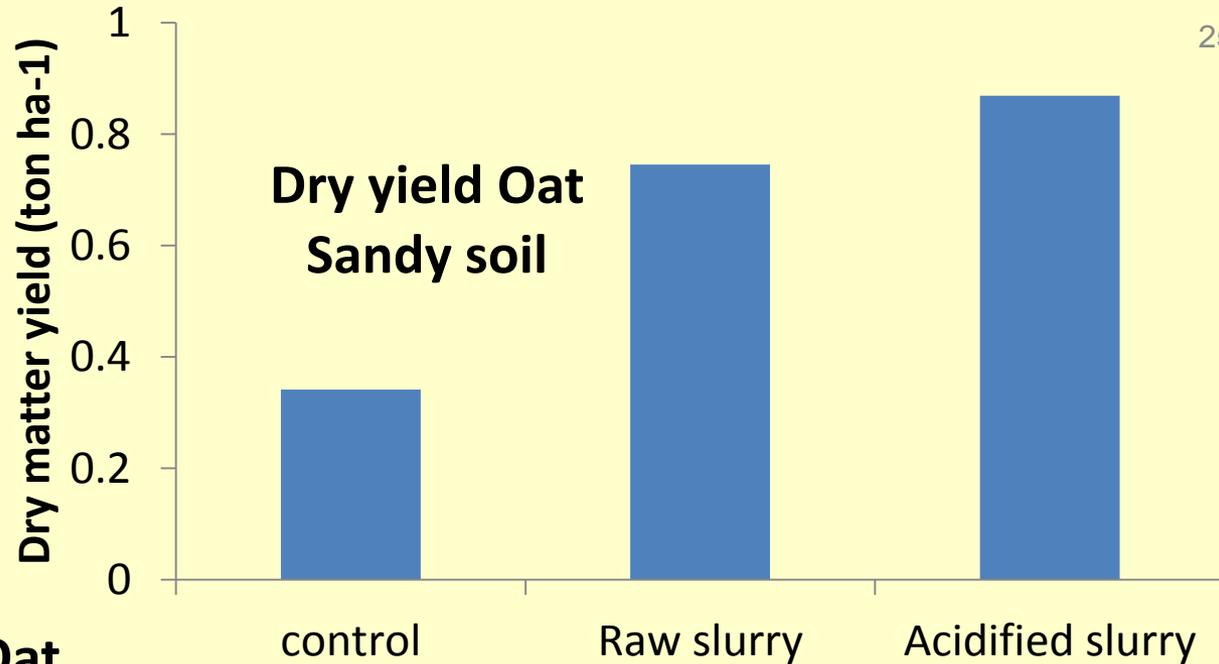
# Agronomic Value

Acidified slurry may act as a starter for maize (Petersen et al., 2013)



# Agronomic Value

25



# Conclusions

<b>Slurry composition</b>	<b>Soluble P</b>	↗
	<b>Inorganic C</b>	↘
<b>Storage</b>	<b>CO<sub>2</sub></b>	→ (initial burst)
	<b>CH<sub>4</sub></b>	↘
	<b>H<sub>2</sub>S</b>	→ (initial burst)
<b>Soil application</b>	<b>Nitrification</b>	↘ (delay)
	<b>N<sub>2</sub>O</b>	→ ↘
	<b>CO<sub>2</sub></b>	↘
	<b>P availability</b>	↗
	<b>Crop yields</b>	↗

**YES, slurry acidification can be used as  
a slurry management tool in  
Mediterranean countries**

# Acknowledgements

**Co-authors:** S. Surgy,  
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M. Hjorth,  
J. Coutinho



Fundação para a Ciência e a Tecnologia (FCT) financially supported this research through the projects “Animal slurry management: sustainable practices at field scale” (PTDC/AGR-PRO/119428/2010) and (ProjectPEst-OE/AGR/UI0528/2011) and a grant to David Fanguero (SFRH/BPD/84229/2012).



The research leading to these results has received funding from the People Programme (Marie Curie Actions) of the European Union's Seventh Framework Program FP7/2007-2013/ under REA grant agreement n° [289887].

# Thank you for you attention



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